

Energy National Research Programmes 70 and 71

Project

PEM fuel cells





Fuel Cells for Sustainable Mobility

Electric cars can draw their electricity from batteries. Hydrogen, too, is a source of energy. In fuel cells, it can be converted into electricity without CO_2 emissions. Researchers have optimized these fuel cells and are now bringing the outcome to market on a website.



A fuel cell in the laboratory: Swiss companies aim to play an important part in the emerging market of fuel cells and hydrogen technology. *Source:* Adobe Stock





At a glance

- Hydrogen and fuel cells can serve to store electricity for later purposes. Swiss researchers are now improving the efficiency of this technology.
- Improving the transport of hydrogen and oxygen through porous gas diffusion layers can reduce losses within the cells.
- Researchers at the PSI and the ZHAW have therefore studied the processes in the pores on a microscopically small scale. Using the resulting data, they have developed mathematical models they are now marketing for commercial purposes.

Presently, a significant proportion of greenhouse gas emissions are caused by road traffic. Consequently, in order to achieve the climate targets, cars and trucks must forgo fossil fuels and switch to electric power. The electricity required can be provided either by batteries or by on-board hydrogen which is converted into electricity in so-called fuel cells. The only waste product is pure water, i.e. no carbon dioxide, nitrogen oxides or particulate matter. The advantage of fuel cells in buses, trucks and cars, as opposed to batteries, is that vehicles can be refuelled with hydrogen very quickly, as is currently the case with petrol. This eliminates long waiting periods while a battery is being recharged. The high energy density of hydrogen enables vehicles to achieve long ranges. A new value chain in the field of hydrogen and fuel cells, to which researchers from the Zurich University of Applied Sciences ZHAW are also contributing, is currently emerging worldwide.



Micropores within the cell

More specifically, the researchers have studied so-called polymer electrolyte fuel cells (PEFC). In these fuel cells, the reaction components, oxygen and hydrogen, must diffuse through a porous gas diffusion layer. However, at high specific performance, measured by weight, considerable electrical voltage is lost during transport of the reaction components. In order to reveal the conditions under which this happens, the researchers screened the gas diffusion layers of fuel cells using X-ray tomography, an imaging process that enables the visualization of minute structures. This allowed them to establish a connection between transportability and properties such as pore distribution, the resistance of narrow pores to transport processes, and non-circular pores. In doing so, they were able to quantify the transportability of different gas diffusion layers. The scientists then compared the X-ray measurements with computer simulations and developed mathematical models that allow fuel cell developers to build more efficient devices. One possible application is the development of new materials with optimized pore designs for gas diffusion layers. The researchers are now in the process of setting up a web platform to market the computer models, so that, in the future, all can benefit from this research work.

How does a fuel cell work?

Fuel cells can be used to technically exploit the energy stored in hydrogen. In these cells, hydrogen and oxygen are brought together under controlled conditions. A chemical reaction then produces electricity and, as a waste product, water.

The electricity is generated because hydrogen and oxygen react with each other (the so-called detonating gas reaction is well known). Hydrogen and oxygen are supplied to the fuel cell separately, allowing the chemical reaction to occur under controlled conditions rather than explosively. Gas diffusion layers are used to achieve an even distribution of the gases on the electrochemically active surfaces. In devices known as gas diffusion electrodes, hydrogen is split into an H⁺ and a negatively charged electron on the anode side. On the cathode side, oxygen is also reduced. Only positively charged particles can pass through a proton-conducting membrane between the two cell sides. The negatively charged electrons generated from the hydrogen must find another way to meet the oxygen on the other side of the barrier and then react to form water. This alternative route is an electrical conductor, in which the flow of electrons generates an electric current. The resulting electricity can then be used to power various devices.



Produkte aus diesem Projekt

- Microstructure-property relationships in a gas diffusion layer (GDL) for Polymer Electrolyte Fuel Cells, Part I: effect of compression and anisotropy of dry GDL Date of publication: 01.01.18
- Microstructure-property relationships in a gas diffusion layer (GDL) for Polymer Electrolyte Fuel Cells, Part II: pressure-induced water injection and liquid permeability Date of publication: 01.01.18
- Modeling the Effects of Using Gas Diffusion Layers With Patterned Wettability for Advanced Water Management in Proton Exchange Membrane Fuel Cells Date of publication: 01.01.18
- An ensemble Monte Carlo simulation study of water distribution in porous gas diffusion layers for proton exchange membrane fuel cells
 Date of publication: 01.01.18
- Predicting effective conductivities based on geometric microstructure characteristics
 Date of publication: 01.01.18
- Characterization of Liquid Water Saturation in Gas Diffusion Layers by X-Ray Tomographic Microscopy Date of publication: 01.01.18
- Free open reference implementation of a two-phase PEM fuel cell model Date of publication: 01.01.18
- Improvements of gas transport characterization in GDL based on XTM imaging Date of publication: 01.01.18
- o Ensemble-based study of

- Numerical simulation of liquid water saturation in cathode side gas diffusion layers of PEFCs Date of publication: 01.01.18
- From pore-scale material properties to macro-homogeneous PEFC modeling Date of publication: 01.01.18
- Numerical modeling of patterned porous materials for thermo-neutral fuel cells
 Date of publication: 01.01.18
- Analysis and Extension of a PEMFC Model
 Date of publication: 01.01.18
- An open implementation of a twophase PEMFC model in MATLAB Date of publication: 01.01.18
- A new open-source PEMFC simulation tool for easy assessment of material parameterizations Date of publication: 01.01.18
- Investigation of liquid water transport in GDLs using X-ray tomographic microscopy Date of publication: 01.01.18
- Characterization of liquid water invasion in GDLs using X-ray Tomographic Microscopy Date of publication: 01.01.18
- Characterization of liquid water invasion in GDLs using X-ray Tomographic Microscopy Date of publication: 01.01.18
- Improvements of gas transport characterization in GDL based on XTM imaging Date of publication: 01.01.18
- o Segmentation of Low Quality Gas



equilibrium liquid water distribution in PEFC gas diffusion layers Date of publication: 01.01.18

- A novel Monte Carlo technique for simulating liquid water distribution in gas diffusion layers of PEFCs Date of publication: 01.01.18
- Voxel-based modelling of water distribution in PEFC porous media Date of publication: 01.01.18
- Parameterisation of macrohomogeneous models of proton exchange membrane fuel cells
 Date of publication: 01.01.18
- Toward predictive PEFC simulation: The importance of thermal and electrical contact resistance Date of publication: 01.01.18
- Advances in Multi-Scale Modeling of PEFCs
 Date of publication: 01.01.18
- Influence of pore scale material properties on the performance of proton exchange membrane fuel cells
 Date of publication: 01.01.18
- Current Challenges in Two-Phase PEMFC Modeling Date of publication: 01.01.18

Diffusion Layer X-ray Tomographic Microscopy Images Date of publication: 01.01.18

- Hydrogen for Electromobility: A Promising Energy Carrier Date of publication: 01.01.18
- Elektromobilität mit Brennstoffzellen Date of publication: 01.01.18
- Approaches and Challenges of Multi-Scale Modeling of Polymer Electrolyte Fuel Cells
 Date of publication: 01.01.18
- Web page on modeling and simulation of electrochemical cells Date of publication: 01.01.18



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